

## Hind Photostat & Book Store

Best Quality Classroom Topper Hand Written Notes to Crack GATE, IES, PSU's & Other Government Competitive/ Entrance Exams

## MADE EASY ELECTRICAL ENGINEERING Material Science By.Suneel Tiwari Sir

- Theory
- Explanation
- Derivation
- Example
- Shortcuts
- Previous Years Question With Solution

## visit us:-www.hindphotostat.com

Courier Facility All Over India (DTDC & INDIA POST) Mob-9311989030 21/12/2023

MATERIAL SCIENCE

## <u>TECHNICAL</u> PAPER-II

# Syllabus. 1) Atomic Structure and Chemical Bonding (618 = Notes page Nor3) 2) Crystallography (618 = Notes page No-21) 3) Insulators / Dielectrics 4) Magnetic Properties (618 = Notes page No-82)

6 Conductors (45 & Notes page No - 33)

© Super Conductors

() Semiconductors (618 & Notes page No-60)

Ceramics / Composites (us à notes page No-97)
Nanomaterials.

Topics 1,2,4,5,7,8 GS-Part में पढामा है। Topics 7 (semiconductors) Electronic वाले जन्में के पढामा है। Topics 4,5 में बुद्द point और Add करवास्मे

Refer Books

D Electrical Engineering Matorials by S.P. Seth ipple 3, 4, 5, 6

3 INSULATORS/DIELECTRICS E → भोद charge किसी Body & Uniformly distributed हैं तो उसका Effect En Centre & Consider कर सकते हैं। -zq € †29 +29 इस की Polouisation कहते है। यह Metal में नहीं होगा नमोन्स वहा  $N = \frac{n}{V_{A}} = \frac{N_{A}g}{A}$ TR free Electron En 21 -> Insulator are generally made of PVC or Nylon. -> Ansulation अमेर Polarisation में दोनों रुम साथ- पाहिर Charge Store कार्न के लिए Dielectrics की जस्त होगी। -> Non-Conducting Material colo not have Sufficient free Electron to take part in Electrical Conductivity But. These material have abundance of bounded Electrons clue to which they get polarised on the application of Electric field.

3

()> The Material which get Easily polarized Under the Influence of Electric field are known as sietectries. -> If the main function of Non-Conducting Material is to provide Electrical Insulation than material is known as Insulabors. -> If the main function of Non-Conducting Material is Storage Charge then it is known as Dielectrics. # <u>Capacitance</u> Consider a parallel plate apacitor.  $C = \frac{\varepsilon_{\rm r}\varepsilon_{\rm o}A}{\varepsilon_{\rm r}} = \frac{\varepsilon_{\rm A}}{\varepsilon_{\rm r}}$ C-> Capacitance (in Farad) A > Cross Section carea of the plates d -> Distance b/w the plates E -> Permittivity of free space or vaccum E = 0.854 X 10-12 F/M

 $(\mathcal{E}_{r} = \mathcal{E}_{r})$  $\mathcal{E}_{\tau} = \operatorname{Relative}$  permittivity or Dielectric Constant of Material.  $\mathcal{E} = \operatorname{Permittivity}$  of Material  $\Rightarrow \mathcal{E} = \mathcal{E}_{\tau} \mathcal{E}_{o}$ 

#_Material	Er	
Vacuum	1	
Ar	1.0006	
Helívm	1.0000684	<b>£</b>
paper	2.0-3.0	
Teflon	2.1	
Fused Quartz	3.8	
Nylon	3.5	
Bakelite	4.9	
Mica	6	
Distilled Water	01	
TCO2	100	
-> For a Capacitor	· · · · · · · · · · · · · · · · · · ·	
Bar		
Q = CV	, ·	
-> Energy stored i	in a Capacitor	
$\mathcal{W} = \frac{1}{2}CV^2 =$	$\frac{1}{2}\frac{\partial^2}{\partial c} = \frac{1}{2}\partial \cdot V$	
-> Energy Density		
	red per Unit Volume	•
	5	

S

6  $D = \mathcal{E}E$  $D \rightarrow \text{Electric flux density } (e/m^2)$  $E \rightarrow \text{Electric field Intensity } (V/m)$  $\mathcal{E} = \frac{D}{F}$ # <u>Electric difse Moment</u> (F) (age 75) → To Equal and opposite charges separated by Certain distance Constitute on Electric colipste. +9 × - 9 Electric collipole Moment (F)  $\vec{p} = q \cdot \vec{d}$ Unit -> O C-m (2) Debye  $1 \text{ Debye} = 3.33 \times 10^{-80} \text{ c-m}$  $\vec{p} \rightarrow \text{St}$  is a vector sugnify and vector collinection is taken from negative charge to positive charge. 31JIT Atomic no. 7 stable 2 of Atom -> 3TIR Molecule no. 7 stable 2 FT Molecule

6

# Polarisation (P) (Cis 21 - 76) -> It is defined as Electric clipste moment per Unit Volume.  $\rightarrow \vec{P} = \frac{\mathcal{E}|ectric \ olipoli Moment}{Volume}$ - On a Macro-Scopic Scale N→NO. of Electric dipole per Unit Volume. F → Electric dipole moment of Each dipole. Acc. to G.S part Then  $\vec{p} = \vec{p} + \vec{p} + - - N + imes$ P=ND N -> No. of Molecules per Unit Volume N. AV -> No. of Molecules in Volume AV.  $\vec{p} = \frac{1}{\Delta V} \sum_{i=1}^{N \cdot \Delta V} \vec{p}_i = N \vec{p}$ p → Electric dipole Moment of Each Molecule # Unit of P - $= \frac{1}{m^3} \times C - m$  $= \frac{C}{m^2}$ 

7

BDR EE -> Total. Electric flux colensity in a colielectric material is take to  $D \rightarrow Applied Electric field (E, E)$   $D \rightarrow Polarisation Inside the Material(P)$ i.e.,  $\vec{D} = \mathcal{E}\vec{E} + \vec{P} - O$  (Ace to Material Science) but  $\vec{D} = \mathcal{E}_{T}\mathcal{E}_{T}\vec{E} - (2)$  (Acc. to Physics ) from Egn () 7 Egn (2) P+EE = EEE  $P = \varepsilon_r \varepsilon_r \vec{E} - \varepsilon_r \vec{E}$  $\vec{P} = \mathcal{E}_{0}(\mathcal{E}_{0} - 1)\vec{P}$ P=ExcE where  $\chi_{p} = E_{r} - 1$ is Electric Susceptibility of Material  $\mathcal{H}_{e} = \frac{P}{\mathcal{E}_{o} \vec{E}} = \frac{Boond}{Free} \frac{Charge}{Charge} \frac{Charge}{Charge$ Free charge density